

The Analysis of β -cryptoxanthin and Zeaxanthin using HPLC in the Accumulation of Orange Color on Lowland Citrus

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Abstract

Citrus peel color is one of the main quality attributes which was caused by the accumulation of carotenoids and its derivatives, especially β -citraurine. It makes citrus peel color looks attractive (orange). The orange color is a mixture of β -cryptoxanthin with β -citraurin. The objectives of this study were (1) to observe the effect of precooling and duration of proper ethylene exposure in the formation of orange color on citrus peel, (2) to identify and determine the β -cryptoxanthin content and total chlorophyll on citrus peel. Citrus was from Tuban, East Java while the study was conducted at PKHT IPB and LIPI. Precooling and without precooling treatment prior to injection of 100 ppm of ethylene exposed at 15 °C, duration of exposure control (0), 24, and 48 hours. The results show that the best color of the Citrus Color Index (CCI) is the precooling treatment and the duration of ethylene exposure for 24 hours, which can reduce total chlorophyll content about 8 times and proved to increase β -cryptoxanthin pigment content five times in accelerating the formation of orange *citrus reticulata* peel color to bright orange. Degreening has no significant effect on total dissolved solids and the firmness level of citrus fruits.

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Keyword

β -cryptoxanthin

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Introduction

Peel color is one of external quality attributes of citrus and plays important factor in customer decision in purchasing. Green-colored citrus is usually considered sour-tasted and unripe, while orange-colored citrus is considered otherwise. Generally, consumers choose orange-colored citrus. Therefore, it is essential to ensure that the citrus fulfills the criteria when it is ready to enter the market.

In order to fulfill the consumers' demand regarding the taste, the accumulation of orange-colored on citrus peel through post-harvest treatment should be done. The treatment is degreening with ethylene. Degreening so far is still not succeed to form orange-colored citrus. In fact, it made the citrus to form yellow-colored peel. This result is also unfavorable because consumers consider yellow-colored citrus is in the verge of senescence.

Degreening in tropical region is still not success because β -*citraurin* pigment as the red color formation is not formed. Only yellow color pigment β -*cryptoxanthin* is formed. (Kato *et al.*, 2006; Fanciullino *et al.*, 2008). Yellow on citrus peel is caused by carotenoids derivative *cryptoxanthin* (Zhou *et al.*, 2010). Orange color can be formed due to the mixture of β -*cryptoxanthin* and β -*citraurine* (Stewart & Wheaton, 1971; Rodrigo *et al.*, 2013). While β -*cryptoxanthin* and *zeaxanthin* is the pigment contributed in the formation of β -*citraurin* (Ma *et al.*, 2013).

The failure of degreening in Indonesia is because the citrus is not exposed to low temperature during its growth. Thus, β -*citraurin* pigment can not be formed. It is only formed at 18-24 °C. To substitute the existence of low temperature, the precooling treatment was introduced. Precooling after harvest is expected to substitute low temperature treatment which could not be fulfilled when the fruit is in the field. Ladaniya (2008) reported that low temperature affects greatly on the color changes on peel.

According to Matsumoto *et al.*, (2009) research about Satsuma fruit storage at 30 °C was proven to increase *zeaxanthin* and β -*carotene* but no β -*citraurin* pigment was found. The main reason of failure was that high temperature prevents the accumulation of β -*citraurin* and reduced orange color (Stewart & Wheaton, 1971). High temperature on tropical regions limits the accumulation of important pigments on citrus and prevents its peel to achieve more favorable color.

Therefore, degreening technology which could induce citrus peel orange, reduce chlorophyll content, and identify β -*cryptoxanthin* as orange formation is needed. The research aims to observe the effect of precooling and obtain the proper ethylene exposure duration and to identify and determine β -*cryptoxanthin* and *Chlorophyll* content on citrus peel before and after degreening.

Materials and Methods

The research was done at PKHT IPB and the analysis of chemical content of peel was done at LIPI Cibinong. It was conducted from January 2016 until July 2017. Materials used in the research were Citrus reticulata "Tejakula", ethylene gas, and liquid nitrogen. Tools used were degreening box, cool storage, syringe, refractometer, color reader, spectrophotometer, and chromatography, HPLC (High Performance Liquid Chromatography). A 2 factors Randomized Group Design was used as research design. Treatments were harvest and exposure duration. Harvest factor consists of 2 levels that is without precooling and precooling. Exposure duration factor consists of 3 levels, without ethylene exposure, exposure for 24 hours, and exposure for 48 hours. There were 6 combinations repeated 3 times and 18 units of experiment.

Citrus reticulata "Tejakula" came from low land estate in Singgahan, Tuban, East Java (40 m asl). After the citrus was harvested and treated, it was brought to PKHT lab, Bogor. It was sorted and put into degreening box and sealed tightly (air tight). Ethylene gas 100 ppm was injected to the degreening box using 5 ml syringe, then the citrus was put in cool storage at 15 °C. The fan was activated during degreening, which is control, 24 hours, and 48 hours.

Hue Angle and Citrus Color Indeks (CCI)

Changes in the color quality of citrus peel was measured with Minolta Color Reader type 310. This tool color notation system is hunter notation system (color system L, a, and b). The measurement results were expressed in the Citrus Color Index (CCI). Hunter color notation system is characterized by three parameters of color, the brightness symbolized by L, the chromatic colors symbolized by 'a' notation and color intensity with the notation 'b'. Notation 'L' states the parameter of brightness (lightness), with value L: value 0 means black and 100 means white. While 'a' notation stating chromatic colors red-green mix with +a (positive) from 0 to +100 indicating red color and value (negative) from 0 to -80 indicating green color, while the notation 'b' states chromatic color, the mixture of blue-yellow value +b (positive) from 0 to +70 indicating yellow color and the -b (negative) value from 0 to -70 indicating blue color (Andarwulan *et al.*, 2011).

The value of Citrus Color Index (CCI) is a formula widely used to see the quality of citrus peel color. According to Jimenez-Cueata *et al.*, (1981), Value Citrus Color Index (CCI) is a formula that is widely used to see the quality of orange peel color. Based on Jimenez-Cueata *et al.*, (1981) that the value of CCI is used as follows the following equation:

$$CCI = \frac{1000 \cdot a}{L \cdot b}$$

Citrus Color Index (CCI) using following range: $CCI \leq -5$ (dark green), $-5 < CCI \leq 0$ (green), $0 < CCI \leq 3$ (yellowish green), $3 < CCI \leq 6$ (greenish yellow), $6 < CCI \leq 8$ (yellowish orange), $8 < CCI \leq 10$ (orange), dan $CCI > 10$ (dark orange).

Total Soluble Solid

The fruit flesh of several samples from each treatment were taken and the total soluble solids was measured using a hand refractometer. A drop of juice was placed on hand refractometer prism.

Fruit Firmness

Fruit firmness is determined by fruit resistance to puncturing the peel done by penetrometer.

Total Chlorophyll

The total content of chlorophyll was measured using spectrophotometry method. Citrus peel was weighed at 0.5 grams and then crushed (slurry) and extracted with 2 mL acetris. The extract then inserted into microtube and centrifuged for 10 seconds. The centrifuged filtrate was inserted in a 1-mL reaction tube, and then 3 mL acetris was added and placed in cuvet to be measured using a spectrophotometer at wavelengths 537, 647 and 663 nm. According to Sims and Gamon (2002), after obtaining an absorbance value, the total content of chlorophyll can be calculated by the following equation:

$$Chl_a = 0.01373 \cdot A_{663} - 0.000897 \cdot A_{537} - 0.003046 \cdot A_{647}$$

$$Chl_b = 0.02405 \cdot A_{647} - 0.004305 \cdot A_{537} - 0.005507 \cdot A_{663}$$

$$Total\ Klorofil = Chl_a + Chl_b$$

Identification of β -cryptoxanthin and zeaxanthin Pigments

Material Preparation

Citrus peels are separated from the edible portion, then immediately cooled with nitrogen liquid in the dewar tube until next step.

Carotenoid Extraction

The identification of carotenoids was performed by the method described by Kato *et al* (2004). The pigment was extracted from the sample with hexane solution: acetone: ethanol (2: 1: 1, v / v) containing 0.1% (w / v) 2,6-ditert butyl 4-methylphenol and 10% (w / v) magnesium carbonate. After the organic solvent evaporates entirely, the carotenoid-containing extract, which is esterified into fatty acid, is disacored with 20% (w / v) methanolic KOH. Water-soluble extracts are removed by saturated NaCl. The pigments were partitioned in the diethyl ether phase and evaporated.

Analysis of β -cryptoxanthin, zeaxanthin and β -carotene

The residue is redissolved in methyl tert butyl ether solution: methanol (4: 6, v / v). An aliquot (30 μ L) is separated by an HPLC reversed phase (Shimadzu, Ascentis) equipped with type C-18, at a 1 mL min⁻¹ flow rate. The standards of β -cryptoxanthin, zeaxanthin and β -carotene were prepared based on the method described with fresh weight micrograms per gram (preparing for calibration of the standard). Quantification of β -cryptoxanthin, zeaxanthin and β -carotene was performed in 3 replications. The data were obtained from average HPLC values and calculated using the calibration equation. With standard pigments (Figure 1), and the calibration formula are as follows (Table 1):

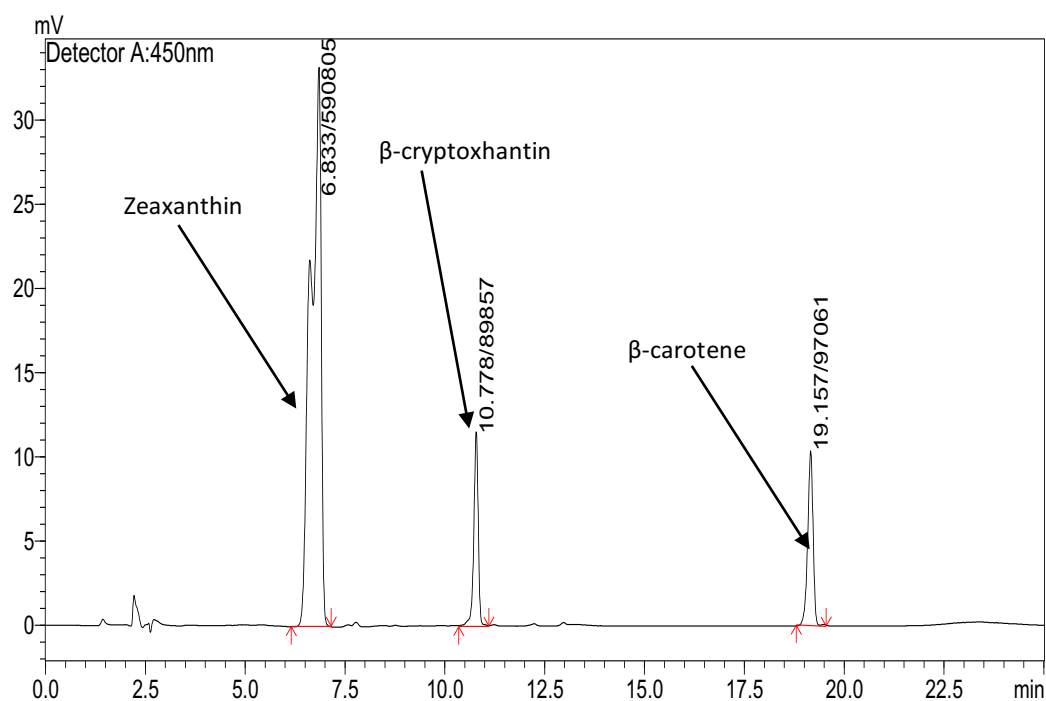


Figure 1. HPLC chromatogram for mixture of authentic sample of Zeaxanthin β -cryptoxanthin, and β -carotene.

Table 1. The linier equation for β -carotene, β -cryptoxanthin, and zeaxanthin pigments

Standard

	Zeaxantin	Criptoxantin	Caroten
Linier equation	$y = 86079x + 44977$	$y = 68174x + 111876$	$y = 3915.3x + 249036$
R2	$R^2 = 0.9938$	$R^2 = 0.9673$	$R^2 = 0.9838$

Data Analysis

The results from observation were tested by using SAS software (Statistical Analysis System) The data were analyzed using Analysis of Variance at 5% significance level. When the result showed significant effect of the treatment, it will be tested further using Duncan Multiple Range Test (DMRT) at 5% level.

Results and discussion

The effect of precooling and exposure duration to *Citrus reticulata* of Tuban origin showed the best color change was on the duration of 24 hours exposure with precooling and degreening treatment. Citrus peel color with precooling treatment on 24-hour ethylene exposure resulted in orange peel color, whereas without precooling showed a yellowish orange color (Figure 2, 3).

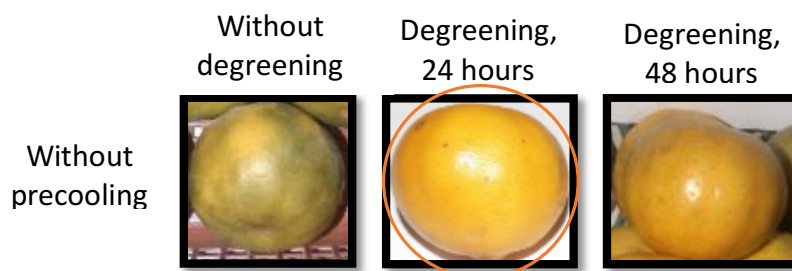


Figure 2. The change of *Citrus reticulata* 'tejakula' peel color without precooling treatment on several ethylene exposure durations (9 days after treatment)

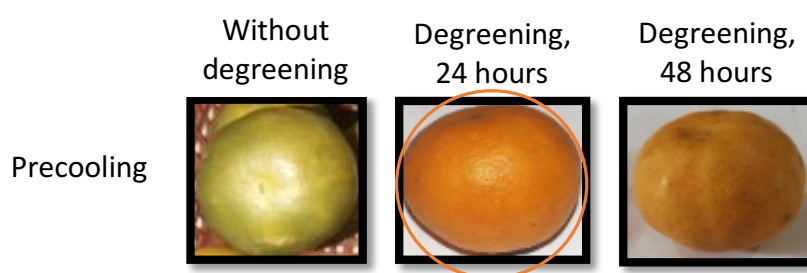


Figure 3. The change of *Citrus reticulata* 'tejakula' peel color with precooling treatment on several ethylene exposure durations (9 days after treatment)

Total soluble solids and and citrus firmness

The measurement for measuring citrus sweetness level is expressed in the content of total soluble solids. Based on the precooling treatment and ethylene exposure duration at the final observation, it was shown that the treatments gave no significant effects to total soluble solids and citrus firmness.

Table 2. The change of *Citrus reticulata* "tejakula' peel color with degreening treatment on several ethylene exposure durations (0, 12 days after treatment)

Treatment	Total soluble solids (° brix)	Citrus firmness (Kgf)
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	0	12	0	12
	Day after treatment			
Precooling				
- Non precooling	10.76	10.76	0.58	0.62
- Precooling	10.77	10.72	0.56	0.62
Exposure durations				
- 0 hour	10.77	10.73	0.57	0.63
- 24 hours	10.75	10.76	0.57	0.63
- 48 hours	10.79	10.74	0.57	0.60

Citrus Color Index Value (CCI)

The use of the Citrus Color Index (CCI) value described colors such as red, orange or green. The increase of CCI value indicates that there is a change of color from green to orange. Precooling with degreening had an effect on the CCI value on *Citrus reticulata* to form orange color faster than treatment without precooling. The result of CCI measurements of *Citrus reticulata* with precooling treatment with degreening on 24-hour exposure duration, it could form orange color on day-9 after degreening with value 7.94. While the CCI value of *Citrus reticulata* given treatment without precooling on 24-hour exposure duration could form an orange color on day-12 of treatment with value 7.99 (data not shown).

Total Chlorophyll

The color variation of citrus fruits is caused by the main groups of chlorophyll pigments and carotenoids. The dominant pigment on unripe or ripe green fruit peel is chlorophyll (Rodrigo *et al.* 2013). The total chlorophyll of *Citrus reticulata* "tejakula" was observed at the time before degreening treatment (harvest, 0 days), without degreening (9 days after treatment) and precooling treatment with degreening (9 days after treatment). The total chlorophyll in *Citrus reticulata* "tejakula" with precooling and degreening treatment with 24 hours exposure decreased sharply with the lowest value compared to other treatments, with a value of 9.81 $\mu\text{g g FW}^{-1}$ on day-9 after treatment, initial chlorophyll content value before treatment 113.81 $\mu\text{g g FW}^{-1}$ (Figure 4).

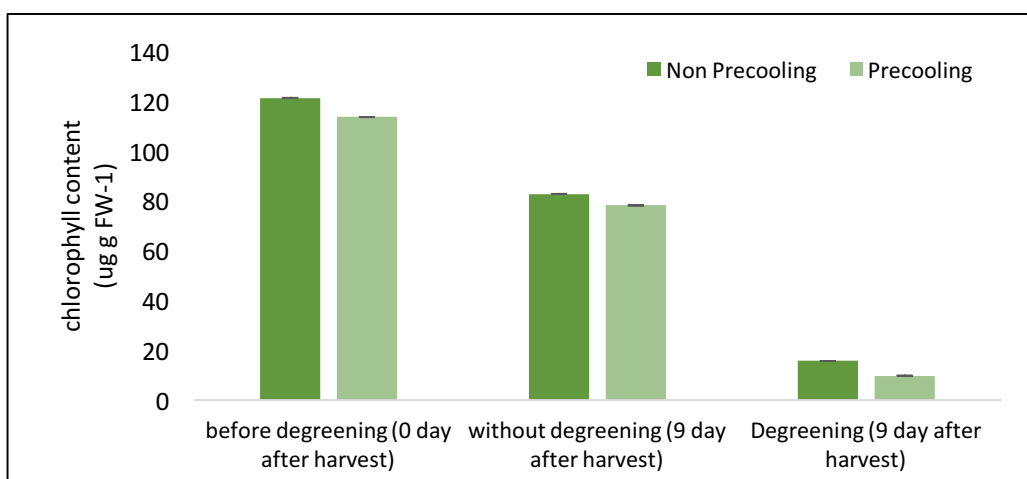


Figure 4. The change of total chlorophyll on citrus peel

Chlorophyll degradation causes color changes during degreening and the appearance of carotenoid pigment to produce orange color. These results are also similar to the study conducted by Stewart and Wheaton (1972); Rodrigo and Zacarias (2007) that ethylene

applications triggered skin pigmentation by stimulating chlorophyll degradation and accumulation of carotenoids, including β -citraurin.

Identification of β -Carotene, β -Ccriptoxanthin and Zeaxanthin

From the result, we can conclude that in order to form orange color on *Citrus reticulata* "Tejakula" in lowland, precooling prior to degreening is needed. Precooling and degreening treatments provided better and faster result in orange color accumulation. HPLC curve on zeaxanthin and β -criptoxanthin pigments (Figure 5). The content of β -criptoxanthin on *Citrus reticulata* "Tejakula" with precooling and degreening treatments increased rapidly around 22.03 ug g FW-1 on day 9 after treatment (Figure 5).

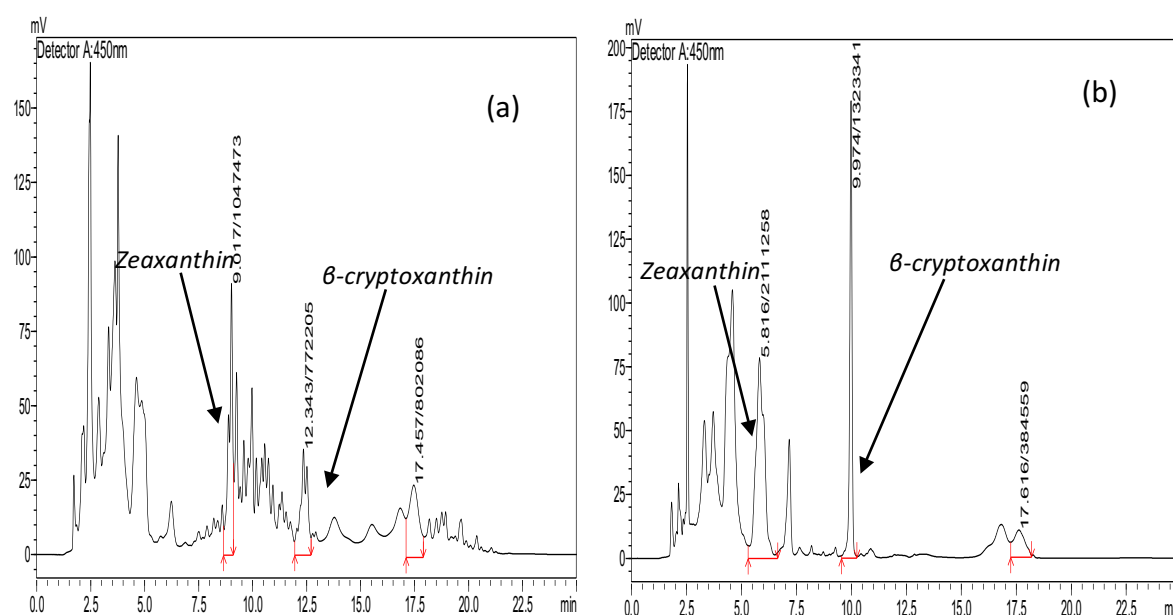


Figure 5. HPLC chromatogram of *Citrus reticulata* "Tejakula" without precooling (a), and precooling (b) (day 9 after treatment) with degreening

The content of zeaxanthin on precooling and degreening treatments and exposure duration for 24 hours resulted in 37.65 ug g FW-1 on day 9 after treatment. Zeaxanthin content without degreening shows lower value that is 12.22 ug g FW-1 (data not shown). This result is in accordance with Kato *et al* (2004); Rodrigo *et al* (2013) who stated that the change of citrus peel color from green to orange caused by nonphotosynthetic carotenoid synthesis, that is β -citraurin acts as the color formation to red-orange on *Citrus nobilis* peel. The accumulation of this compound is determined by precursor availability in carotenoids which is photosynthetic such as zeaxanthin, β -criptoxanthin, karotenoid nonphotosintetic (β -citraurin) which only take place in low temperature.

Conclusions

The Citrus Color Index (CCI) is the precooling treatment and the duration of ethylene exposure for 24 hours, which can reduce total chlorophyll content about 8 times and proved to increase β -criptoxanthin pigment content five times in accelerating the formation of orange *Citrus reticulata* peel color to bright orange. Degreening has no significant effect on total dissolved solids and the firmness level of citrus fruits.

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